# SUMMATIVE ASSESSMENT - I - 2016 GENERAL SCIENCE - Paper - I

(Physical Sciences)

(English Medium)

PART-A&B

Max. Marks: 40

Time: 2.45 Mts.

#### Instructions:

Class : X

- 1. In the time duration 2H 45 Mts; 15 Minutes of time is allotted to read and understand the question paper.
- 2. Answer the question paper under Part A on separate answer book.
- 3. Write the answers to the questions under part B on the question paper itself and attach it to the answer book of part A.

Marks: 30

PART - A

#### SECTION - I

 $(4 \times 1 = 4)$ 

Note:

- 1) Answer all the questions.
- 2) Each question carries ONE mark.
- 3) Answer each question in 1 or 2 sentences.
- 1. Why copper bottomed vessels are used for cooking?
- 2. What is the use of keeping food in air tight containers?
- 3. What are the precautions to be taken while diluting an acid?
- 4. Write the lens makers formula.

## **SECTION - II**

 $(5 \times 2 = 10)$ 

Note:

- 1) Answer all the questions.
- 2) Each question carries Two marks.
- 5. Write any two differences between evaporation and boiling.
- 6. Frame any two questions to explain the concept of rancidity.

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7. The data given in the table shows the object distance (u) and image distance (v) of a real image of a concave mirror. Basing on the data answer the following questions.

S.No	u (cm)	v (cm)
1	15	30
2	20	20
3	30	15

- i) What is the formula to find the focal length of the mirror?
- ii) What is the focal length of the mirror?
- 8. Write any four uses of bleaching powder.
- 9. The absolute refractive index of benzene is 1.5. What is its critical angle?

### SECTION-III

 $(4 \times 4 = 16)$ 

Note:

- 1) Answer all the questions.
- 2) Each question carries FOUR Marks.
- 3) There is an internal Choice for each question.
- 10. (A) How much energy is absorbed when 10g of ice at 0°C becomes steam at 100°C.

(Or)

- (B) The focal length of a converging lens is 20 cm. An object is 60 cm from the lens. Where will be the image formed? Write the nature of the image.
- 11. (A) How do you verify that experimentally sin i/sin r is a constant.

(Or)

- (B) Write an experiment to perform a decomposition reaction using calcium carbonate. How do you test the gas evolved in the reaction?
- 12. (A)  $Fe_2O_3 + 2AI \xrightarrow{\Delta} 2Fe + AI_2O_3$

A teacher says that the above reaction involves different types of chemical reaction such as displacement, endothermic, oxidation, reduction and redox. Justify the above statement with proper explanation.

(Or)

(B) Based on the properties of acids, bases and neutral solutions, fill the following table.

Indicators	Acedic solution	Basic solution	Neutral solution
Red litmus			No change in colour
Blue litmus	Red		
Phenolphthalein	No change in colour		
Methyl orange		Yellow	47
Universal			Parrot green

13. (A) Dentists use concave mirror to observe the teeth. Draw a ray diagram which represent the position of teeth and its image. Write the characteristics of the image.

(Or)

(B) Draw a neat diagram, which represent the reaction of zinc granules with dilute hydrochloric acid and mention how do you test for hydrogen gas.

Regd. No.:

60-A

Marks:

SUMMATIVE ASSESSMENT - I - 2016

GENERAL SCIENCE - Paper - I

(Physical Sciences)

(English Medium)

PART - B

Class: X

Max Marks: 10

Academic Standards	AS1	AS2	AS3	AS4	AS5	AS6	Total
Question No.s	1, 4, 5 8,10 14 - 25	2,6 26 27	3,11 28 - 29	7,12	13	30 - 33	33
Marks Allotted	16	4	6	6	4	4	40
Marks Obtained						tena sal	1
Grade					to the	Park se	3-01-,

PART - B

[Marks: 10

## Note:

- i) Answer the following questions.
- ii) Each question carries  $\frac{1}{2}$  Mark.
- iii) Marks will not be awarded in any case of over-writing, rewritten or erased answers.
- iv) Write the capital letter (A, B, C, D) showing the correct answer for the following questions in the brackets provided against them.
  - 14. CGS unit of specific heat is



B)  $\frac{g^{\circ}}{g^{\circ}}$ 

C) can

D) caloc

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	AN	NDHRA PRADESH		
0-/	A materials	2		
15.	When water is boiling, its temp	erature	(	)
	A) Remains constant	B) Increases		
	C) Decreases	D) Both B and C		
16.	Oxidation means	(10.00 (0.0)	(	)
	i) Gain of Oxygen ii) Loss of Hy iii) Loss of Oxygen	drogen iv) Gain of Hydrogen		
	A) i and iv B) iii and iv	C) i and ii D) i, ii, iii, iv		
17.	1 mole of Hydrogen: 0.602 x 10	$0^{24}$ molecules :: 0.5 mole of $H_2$ :		
	molecules.	Ch.	(	)
	A) 0.301 <b>x</b> 10 <sup>12</sup>	B) 6.02 <b>x</b> 10 <sup>23</sup>		
	C) $0.602 \times 10^{12}$	D) $0.301 \times 10^{24}$		
18.	Magnification of a mirror, m =		(	)
	A) $\frac{-v}{u}$ B) $\frac{u}{v}$	C) $\frac{h_o}{h_i}$ D) $\frac{-h_i}{h_o}$		
19.	A: When a mirror is immersed in	n water, its focal length doesn't cha	nge.	
	R: Focal length of a mirror is ind	ependent of surrounding medium.		
		(	)	
	A) A and R are correct and R is	correct explanation of A.		
	B) A and R are correct but R is r	not correct explanation of A.		
	C) A is correct and R is incorrect	t.		
	D) A is incorrect but R is correct	t,		
20.	Match the following.	Anna A . A . A . A . A . A . A . A . A . A	(	1
	A	<u>B</u>		,
	P) Plaster of paris	i) NAHCO <sub>3</sub>		
	Q) Gypsum	ii) CaSO 2H O		

R) Baking soda

S) Washing soda

A) P - iv, Q - ii, R - i, S - iii

C) P - ii, Q - iv, R - i, S - iii

iii) Na<sub>2</sub>CO<sub>3</sub>

iv) CaSO<sub>4</sub>. ½ H<sub>2</sub>O

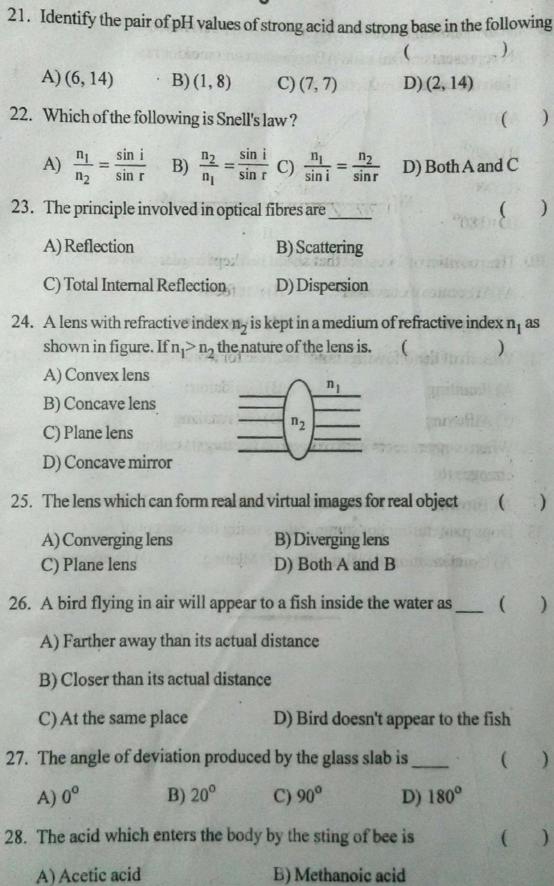
B) P - iv, Q - ii, R - iii, S - i

D) P-ii, Q-iv, R-iii, S-i

[Contd...3

A-UO	6	0	-A
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3



D) Fatty acid

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C) Sulphuric acid

60-A

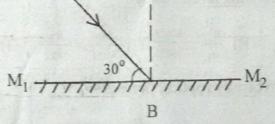
29. In the following figure M<sub>1</sub> M<sub>2</sub> represents a plane mirror, N represents normal and AB represents an incident ray. Then the angle of reflection is

A) 30°

B) 60°

C) 90°

D) 180°



30. The position of a vessel that should be kept in solar cooker

A) At centre of curvature B) At focus

C) At pole

- D) Any where
- 31. Which of the following is not suitable for avoiding corrosion

A) Painting

B) Oxidation

C) Alloying

- D) Galvanizing
- 32. When copper reacts with oxygen on heating its colour changes to

- A) Brown B) Green C) Black D) Yellow
- 33. Dogs pant during hot summer days using the concept of

A) Condensation B) Humidity C) Melting

- D) Evaporation