01. HEAT

- 1. Convert 47°C into Kelvin scale.
- 2. Which factors that effect the rate of evaporation?
- 3. Sheela has confused about Latent heat and specific heat. Can you clarify? How?
- 4. Which of the following substances take more time to raise its temperature for a certain degree celcius. Give reason.

	Kerosene	Water	Sea water
Specific Heat (In cal/gm. °C)	0.50	1.00	0.95

- 5. Why do pigs toil in mud?
- 6. When ice melts, what happened to its temperature?
- 7. The rate of raise in temperature depends upon the nature of the substance. How can you prove it experimentally?
- 8. How much energy is released in which 20 gm of water at 0°C freezes to ice at 0°C?
- 9. If we provide heat to ice at 0°C, why the temperature does no changes until the ice changes to water?
- 10. What are the conditions for thermal equilibrium of two bodies?
- 11. Write the differences between fog and dew.
- 12. If we mix 100ml of water at 90°C to 200ml of water at 60°C. Then find the temperature of the system at thermal equilibrium?
- 13. Your teacher made an experiment that shows the formation of dew and frost. Explain how you show the formation of dew and frost?weebly.com
- 14. Lalitha wants to determine the specific heat of Aluminium shots. What apparatus or material is required to do this experiment?
- 15. How can you prove that the kinetic energy of molecules of a hotter body is greater than that of a colder body?
- 16. Evaporation and condensations are opposite processes. Explain.
- 17. We should not keep a bottle in the deep freezer, which is filled with water up to its brim. Why?
- 18. Water was taken in two glasses. The temperature of water in both glasses is 60°C. Reshma said that they are in thermal equilibrium. Do you agree with Reshma?
- 19. What is the principle of method of mixtures, according to heat?
- 20. What is the effect of temperature on the kinetic energy of the molecules in a substance?
- 21. Give examples for the following processes.
 - (i) Condensation
- (ii) Evaporation
- (iii) Transfer of heat
- (iv) Sublimation
- 22. How do you understand about Warming process? Give two examples.

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