

CLASS-09

PHYSICAL SCIENCE

UNIT PLAN

CHAPTER: 06 – WHAT IS INSIDE THE ATOM ?

| PERIOD NUMBER | CONCEPTS / TEACHING POINTS | PAGES IN TEXT BOOK | | REMARKS |
|---------------|--|--------------------|-----|---------|
| | | FROM | TO | |
| 1. | Atoms Sub atomic particles Electron, proton and neutrons | 89 | 91 | |
| 2. | The structure of an atom Thomson's model of the atom | 91 | 92 | |
| 3. | Rutherford's alpha particles scattering experiment Rutherford's observations Limitations of Rutherford's model of the atom | 92 | 94 | |
| 4. | Bohr's model of an atom Energy levels | 94 | 95 | |
| 5. | Distribution of electrons in different orbits | 95 | 96 | |
| 6. | Valency - concept <small>ignitephysics.weebly.com</small> | 97 | 97 | |
| 7. | Valencies of different elements Electrons arrangement in different atoms | 98 | 98 | |
| 8. | Importance of valency | 98 | 99 | |
| 9. | Atomic number Atomic mass number Writing symbols of atoms | 99 | 99 | |
| 10. | Isotopes Isotopes – Examples Isotopes and their applications | 100 | 101 | |

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